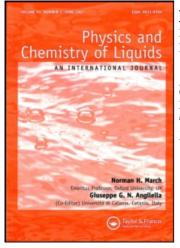
This article was downloaded by: On: *28 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Physics and Chemistry of Liquids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713646857

Structure factors of liquids containing chemical complexes

N. H. March^a; M. Parrinello^b; M. P. Tosi^c ^a Department of Physics, Imperial College, SW ^b Institute of Theoretical Physics, The University, Messina, Italy ^c Guglielmo Marconi Institute of Physics, The University, Rome, Italy

To cite this Article March, N. H., Parrinello, M. and Tosi, M. P.(1976) 'Structure factors of liquids containing chemical complexes', Physics and Chemistry of Liquids, 5: 1, 39 – 43 To link to this Article: DOI: 10.1080/00319107608084105

URL: http://dx.doi.org/10.1080/00319107608084105

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Phys. Chem. Liq., 1976, pp. 39–43 © Gordon and Breach Science Publishers, Ltd. Printed in Dordrecht, Holland

Structure Factors of Liquids Containing Chemical Complexes[†]

N. H. MARCH[‡], M. PARRINELLO[§] and M. P. TOSI

(Received April 15, 1975)

A theory of the three partial structure factors of an A-B alloy containing $A_{\mu}B_{\nu}$ complexes is set up as composites of the structure factors for the three component system. Applying the theory to liquid Ga with Ga₂ complexes, the shoulder observed on the liquid structure factor at 50°C is interpreted as arising from bond-centre (Ga₂)-Ga correlations.

1 INTRODUCTION

Most theoretical treatments of the structure factors of liquid metals and their alloys start out from the assumption of central pair potentials. However, there is compelling evidence from the phase diagrams and other properties of certain binary alloys that chemical complexes can form, even when bonding forces are greatly weakened by metallic screening.

Therefore, in this paper, we present the theory of partial structure factors in A-B alloys in which chemical complexes $A_{\mu}B_{\nu}$ occur with a concentration to be determined from the law of mass action, as discussed by Bhatia and Hargrove.³ We shall do this by calculating the X-ray scattered intensity I(K) from the alloy: the modifications required to pass to neutron scattering are obvious.

2 MODEL OF CHARGE DENSITY AND X-RAY SCATTERING

Following standard practice in the case of crystalline materials, we shall assume that we can associate with A and B nuclei, whether free, or bound

tWork supported by the Gruppo Nazionale di Struttura della Materia del C.N.R. The contribution of one of us (NHM) to this work was mainly carried out during a visit to the University of Rome in the Spring of 1975.

[‡]Department of Physics, Imperial College, SW7

[§] Institute of Theoretical Physics, The University, Messina, Italy

Guglielmo Marconi Institute of Physics, The University, Rome, Italy

in complexes $A_{\mu}B_{\nu}$, localized spherical densities $\rho_{A}(r)$ and $\rho_{B}(r)$. Let the free A atoms be at positions I_{1} , B at I_{2} while the "centres" of $A_{\mu}B_{\nu}$ are at L. Then, measured from L, the μ A's in the complex are at positions \mathbf{R}_{1i}^{L} , i = 1 to μ , the ν B's being at \mathbf{R}_{2j}^{L} , j = 1 to ν . For such a fixed configuration of nuclei, the total electron density is represented by

$$\rho(\mathbf{r}) = \sum_{\mathbf{l}_{1}} \rho_{A}(\mathbf{r} - \mathbf{l}_{1}) + \sum_{\mathbf{l}_{2}} \rho_{B}(\mathbf{r} - \mathbf{l}_{2}) + \sum_{\mathbf{L}} \left[\sum_{i=1}^{\mu} \rho_{A}(\mathbf{r} - \mathbf{L} - \mathbf{R}_{1i}^{L}) + \sum_{j=1}^{\nu} \rho_{B}(\mathbf{r} - \mathbf{L} - \mathbf{R}_{2j}^{L}) \right].$$
(1)

The X-ray scattering factor f(K), given by $\int \rho(\mathbf{r}) \exp(i\mathbf{K} \cdot \mathbf{r}) d\mathbf{r}$ is then obtained in terms of the atomic scattering factors $f_A(K)$ and $f_B(K)$ as

$$f(K) = f_A(K) F_A + f_B(K) F_B$$
⁽²⁾

where

$$F_{\mathbf{A}}(\mathbf{K}) = \sum_{\mathbf{l}_{1}} \exp(\mathbf{i}\mathbf{K} \cdot \mathbf{l}_{1}) + \sum_{\mathbf{L}} \sum_{i=1}^{\mu} \exp(\mathbf{i}\mathbf{K} \cdot \{\mathbf{L} + \mathbf{R}_{1i}^{\mathsf{L}}\})$$
(3)

and

$$F_{\mathbf{B}}(\mathbf{K}) = \sum_{\mathbf{l}_2} \exp(i\mathbf{K} \cdot \mathbf{l}_2) + \sum_{\mathbf{L}} \sum_{j=1}^{\nu} \exp(i\mathbf{K} \cdot \{\mathbf{L} + \mathbf{R}_{2j}^{\mathbf{L}}\}).$$
(4)

Forming the X-ray intensity I(K) as $|f(K)|^2$ and taking the liquid average, the usual partial structure factors of a two-component system are readily identified as

$$\begin{cases} N_{A}S_{AA} = \langle F_{A}F_{A}^{*} \rangle \\ 2(N_{A}N_{B})^{\frac{1}{2}}S_{AB} = \langle F_{A}F_{B}^{*} + \text{complex conjugate} \rangle \\ N_{B}S_{BB} = \langle F_{B}F_{B}^{*} \rangle \end{cases}$$
(5)

where N_A and N_B are the total numbers of A and B atoms in the alloy.

It is evident from Eqs. (3-5) that if we wish to compute these partial structure factors, say in Mg-Bi, near the concentration of the intermetallic compound, then we must have correlations between thel₁'s, l₂'s, the complex centres L and the cross-correlations, as zeroth order input information. At the next stage, angular correlations would need to be considered. However, even "excluded volume" effects have not yet been calculated for a 3-component hard sphere mixture, though we are currently attempting this by the method based on the singularities in the structure factors, which has been developed in another place.⁵ When such information becomes available, the above equations afford a route for estimating the composite structure factors S_{AA} , S_{BB} and S_{AB} , which obviously contain through their

dependence on the concentration of complexes (i.e. on temperature) the binding energy of the complex $A_{\mu}B_{\nu}$. Because this lack of information precludes a calculation on Mg-Bi say, which we regard as the major objective of the present theory, we shall illustrate the method below by discussing pure liquid Ga.

3 ILLUSTRATIVE EXAMPLE: LIQUID GA

While complexes are widely accepted as existing in a material like Mg-Bi, the case of Ga is more open to discussion, but there have been numerous suggestions earlier that Ga₂ complexes can exist in the pure liquid. In this case, with the notation Ga ≈ 1 , Ga₂ ≈ 2 , the above equations immediately lead to the X-ray scattered intensity as

$$I(K) = f^{2}(K) \left[N_{1}S_{11}(K) + 2(N_{1}N_{2})^{\dagger}S_{12}(K) + N_{2}S_{22}(K) \right]$$

= Nf^{2}(K) S(K) (6)

where f(K) is the atomic scattering factor of Ga and, explicitly, in an obvious notation

$$2(\mathbf{N}_1\mathbf{N}_2)^{\frac{1}{2}}\overline{\mathbf{S}_{12}(\mathbf{K})} = \langle \sum_{\mathbf{l},\mathbf{L}} \exp\left(\mathbf{i}\mathbf{K}\cdot\mathbf{l} - \mathbf{L}\right) 2\cos\mathbf{K}\cdot\mathbf{R}_{\mathbf{L}} + \mathrm{c.c.}\rangle$$
(7)

and

$$N_2 \overline{S_{22}(K)} = \langle |\sum_{L} \exp(i\mathbf{K} \cdot \mathbf{L}) 2 \cos \mathbf{K} \cdot \mathbf{R}_{L}|^2 \rangle$$
(8)

where the separation in Ga_2 has been taken to be 2**R** for convenience.

We can safely assume small concentrations of Ga₂ complexes in interpreting the structure data of Ascarelli¹ and then, at least in the interesting region around the first peak of S(K) in Ga the new structural element beyond S₁₁ is S₁₂(K). Assuming for a moment a fixed Ga₂ bond length 2R, S₁₂(K) can be written more explicitly as

$$\overline{\mathbf{S}_{12}(\mathbf{K})} = 2(\rho_1 \rho_2)^{\frac{1}{2}} \int d\mathbf{r} \, d\Omega \left[\mathbf{g}_{12}(\mathbf{r}) - 1 \right] \mathbf{P}(\Omega, \mathbf{r}) \cos \mathbf{K} \cdot \mathbf{r} \cos \mathbf{K} \cdot \mathbf{R}, \tag{9}$$

 ρ_1 and ρ_2 now being number densities while $g_{12}(r)$ is the bond-centre (Ga₂)-Ga correlation function. Finally, in (9), P(Ω , r) is the probability of a solid angle Ω between the axis of a Ga₂ complex and the vector r joining its centre to a free Ga atom. Of course, with relatively weak bonding, we must eventually average over a distribution of R, to account for vibrational fluctuations. If we now define a function

$$\mathbf{F}(\mathbf{K}\mathbf{R},\mathbf{r}) = \int \mathrm{d}\Omega \ \mathbf{P}(\Omega,\mathbf{r}) \cos \mathbf{K} \cdot \mathbf{R}$$
(10)

then to make further progress we must have recourse to models.

The simplest of these, that of free rotation of the Ga₂ complexes, leads to $F(KR, r) = j_0(KR)$, with $j_0(x)$ the spherical Bessel function sinx/x. The important property of this model is that F(KR, r) is independent of r, and hence, from eqs. (9) and (10), $S_{12}(K) \cong j_0(KR) S_{12}(K)$, $S_{12}(K)$ denoting the bond-centre-Ga structure factor. The only information we have on $S_{12}(K)$ which is relevant here is (a) that for two-component hard spheres^{7,2} and (b) experimental data for $Cu_6Sn_5^4$. Though (b) is obviously not for a small concentration of a larger component Sn, it is nevertheless striking that S₁₂ has a pronounced, and narrow, peak at, or very near, the Cu-Cu peak. We stress too that the size differences in Cu-Sn are comparable with those appropriate to Ga_2 -Ga⁺. When we go to the data (a), which is known as a function of concentration, again for small size differences, we find the peak of S_{12} , though less sharp than for Cu-Sn, is again close to the peak in S_{11} , 1 being the smaller component. It is the presence of this peak in $S_{12}(K)$ to which we ascribe the shoulder in S(K) for liquid Ga at 50°C. The factor $j_0(KR)$ multiplying $S_{12}(K)$ will not be reflected in the experiments at all directly, firstly because of the distribution in R which will cause smearing and secondly because the form of $P(\Omega, \mathbf{r})$ in (10) will affect the shape of F(KR, r).

Concerning this latter point, we have also worked out an extreme correlation model in which, to get at $S_{12}(K)$, we assumed that, for a given direction of a Ga₂ axis, a free Ga lay along a line through one nucleus of the Ga₂ complex, making an angle α with the Ga₂ axis. We then obtained the result

$$\overline{S_{12}(K)} = \int_{R}^{R \tan \alpha} 4\pi r^2 [g_{12}(r) - l] [j_0(K|R - r|) + j_0(K|R + r|)].$$
(11)

This model was suggested by the β -phase of crystalline Ga (from a summary of the facts, see for example reference 1), in which there is a zig-zag chain, the angle α being $\cong 72^{\circ}$. Though the result is model dependent, it appears that if we introduce a distribution function in angle α with a sufficiently large width, then the result for $S_{12}(K)$ will again reflect the structure factor $S_{12}(K)$, the Fourier transform of $g_{12}(r)$ in (11).

But we stress that, in each model, the essential point is to correlate three atoms in a manner transcending a simple central pair potential model. In the limit of sufficiently low temperatures, we anticipate that $S_{12}(K)$ will have to reflect the type of correlations in crystalline phases but our models are presently too primitive on this point.

 $^{^{+}}$ To achieve consistency with thermodynamics, i.e. with the K = 0 limit in eqn (6), the Kirkwood-Buff formula for the compressibility of a two-component system requires us to take the volume of Ga₂ to be twice that of G_a.

STRUCTURE FACTORS OF LIQUIDS

4 SUMMARY

In summary, the theory of A–B alloys with compound formation embodied in Eqs. (3–5) can be developed approximately for pure Ga metal for sufficiently small concentrations of the assumed Ga₂ complexes. It is argued from available data on two-component systems that it is the peak in the bond-centre – Ga structure factor which is responsible for the shoulder in S(K) observed at 50 °C by Ascarelli¹. However, and of greater importance, when 3-component hard sphere data becomes available, calculations on an alloy such as Mg–Bi, with Mg₃Bi₂ complexes, will now be possible from the theoretical approach outlined here.[†]

References

- 1. P. Ascarelli, Phys. Rev., 143, 36 (1966).
- 2. N. W. Ashcroft and D. C. Langreth, Phys. Rev., 156, 685 (1967).
- 3. A. B. Bhatia and W. H. Hargrove, Phys. Rev. B10, 3186 (1974).
- 4. J. E. Enderby, D. M. North and P. A. Egelstaff, Phil. Mag., 14, 961 (1966).
- 5. M. Gillan, B. Larsen, M. P. Tosi and N. H. March, 1975 to be published.
- 6. J. G. Kirkwood and F. P. Buff, J. Chem. Phys., 19, 774 (1951).
- 7. J. L. Lebowitz, Phys. Rev., 133, A895 (1964).